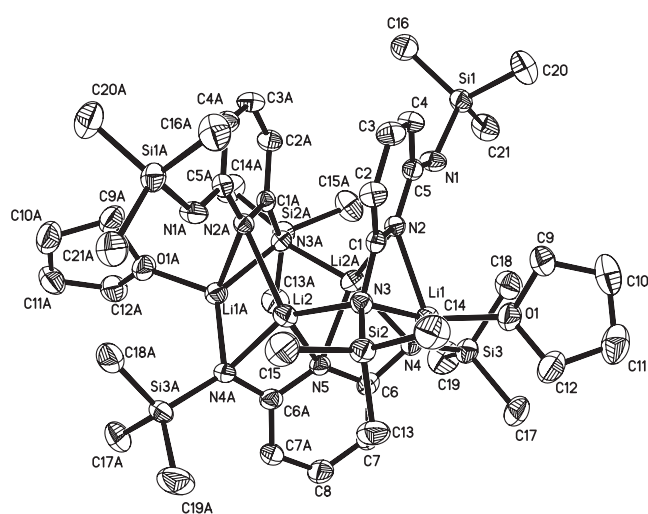


# Crystal structure of bis- $\mu_3$ (*N,N'*-bis-(trimethylsilyl)-pyridine-2-amine-6-amido)- $\mu_4$ (*N,N'*-bis-(trimethylsilyl)-pyridine-2,6-diamido)-di-tetrahydrofuran-tetralithium, [(C<sub>11</sub>H<sub>21</sub>N<sub>3</sub>Si<sub>2</sub>)(C<sub>11</sub>H<sub>22</sub>N<sub>3</sub>Si<sub>2</sub>)<sub>2</sub>(C<sub>4</sub>H<sub>8</sub>O)<sub>2</sub>Li<sub>4</sub>]

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Received January 30, 2008, accepted and available on-line July 28, 2008; CCDC no. 1267/2290



## Abstract

C<sub>41</sub>H<sub>81</sub>Li<sub>4</sub>N<sub>9</sub>O<sub>2</sub>Si<sub>6</sub>, monoclinic, *C*12/*c*1 (no. 15),  
 $a = 24.285(1) \text{ \AA}$ ,  $b = 11.8023(7) \text{ \AA}$ ,  $c = 20.221(1) \text{ \AA}$ ,  
 $\beta = 104.798(5)^\circ$ ,  $V = 5603.5 \text{ \AA}^3$ ,  $Z = 4$ ,  $R_{gt}(F) = 0.043$ ,  
 $wR_{ref}(F^2) = 0.098$ ,  $T = 133 \text{ K}$ .

## Source of material

A solution of two equivalents of *tert*-butyl-(4-methyl-pyridin-2-yl)-amine in THF was treated at  $-50^\circ\text{C}$  with two equivalents of *n*-butyllithium followed by a reaction with one equivalent of CuBr<sub>2</sub> suspended in THF. After removal of the solvent the residue was extracted twice with hexane and filtered. Colourless crystals suitable for X-ray analysis were obtained after slow evaporation of the solvent.

## Experimental details

The H atoms bonded to the N atoms were located and refined. The remaining H atoms were accounted at calculated positions.

## Discussion

To our best knowledge there are no metal complexes known containing the *N,N'*-bis-(trimethylsilyl)-pyridine-2,6-amine ligand in its neutral, deprotonated or double deprotonated form [1,2]. Only a few examples of complexes of a cyclic ligand showing a similarity to the ligand presented here are existing but were not characterized *via* X-ray single crystal structure analysis [3].

The title complex contains one double deprotonated ligand and two single deprotonated ligands coordinating four lithium atoms.

The coordination of each lithium atom is a best described as distorted tetrahedron. The Li2 atom is coordinated by two N<sub>pyridine</sub> atoms and two N<sub>amido</sub> atoms with  $d(\text{Li2}-\text{N}_{\text{amido}}) = 2.014 \text{ \AA}$  or  $2.053 \text{ \AA}$  and  $d(\text{Li2}-\text{N}_{\text{pyridine}}) = 2.109 \text{ \AA}$  or  $2.287 \text{ \AA}$ . The Li1 atom is coordinated by two N<sub>amido</sub> atoms, one N<sub>pyridine</sub> atom and one O atom with  $d(\text{Li1}-\text{N}_{\text{amido}}) = 2.054 \text{ \AA}$  or  $2.086 \text{ \AA}$ ;  $d(\text{Li1}-\text{N}_{\text{pyridine}}) = 2.179 \text{ \AA}$  and  $d(\text{Li1}-\text{O}) = 1.931 \text{ \AA}$ .

**Table 1.** Data collection and handling.

Crystal:	colorless prism, size $0.31 \times 0.40 \times 0.75 \text{ mm}$
Wavelength:	Mo $K\alpha$ radiation ( $0.71073 \text{ \AA}$ )
$\mu$ :	$1.88 \text{ cm}^{-1}$
Diffractometer, scan mode:	STOE-IPDSII, $\omega$
$2\theta_{\text{max}}$ :	$51.52^\circ$
$N(hkl)_{\text{measured}}$ , $N(hkl)_{\text{unique}}$ :	35770, 5312
Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ :	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$ , 3954
$N(\text{param})_{\text{refined}}$ :	294
Programs:	SIR-97 [4], SHELXL-97 [5]

**Table 2.** Atomic coordinates and displacement parameters (in  $\text{\AA}^2$ ).

Atom	Site	$x$	$y$	$z$	$U_{\text{iso}}$
H(2)	8f	0.4279	0.5037	0.8506	0.035
H(3)	8f	0.4842	0.6630	0.8544	0.038
H(4)	8f	0.5779	0.6467	0.8459	0.035
H(7)	8f	0.575	-0.1244	0.8221	0.047
H(8)	4e	$\frac{1}{2}$	-0.2216	$\frac{3}{4}$	0.052
H(9A)	8f	0.5546	0.3982	1.0158	0.056
H(9B)	8f	0.6017	0.3944	0.9724	0.056
H(10A)	8f	0.6323	0.3761	1.1048	0.066
H(10B)	8f	0.6676	0.3114	1.0585	0.066
H(11A)	8f	0.6400	0.1478	1.0944	0.069
H(11B)	8f	0.5987	0.2126	1.1328	0.069
H(12A)	8f	0.5605	0.0938	1.0193	0.057
H(12B)	8f	0.5231	0.1888	1.0450	0.057
H(13A)	8f	0.4067	0.0627	0.8380	0.062
H(13B)	8f	0.3790	0.0745	0.9014	0.062
H(13C)	8f	0.4467	0.0800	0.9135	0.062
H(14A)	8f	0.4497	0.3241	0.9821	0.071
H(14B)	8f	0.3857	0.2859	0.9780	0.071
H(14C)	8f	0.3977	0.4051	0.9466	0.071
H(15A)	8f	0.3311	0.3721	0.8045	0.073
H(15B)	8f	0.3044	0.2662	0.8348	0.073
H(15C)	8f	0.3292	0.2503	0.7692	0.073
H(16A)	8f	0.6743	0.6067	0.7256	0.067
H(16B)	8f	0.7008	0.7106	0.7743	0.067
H(16C)	8f	0.6342	0.6830	0.7592	0.067
H(17A)	8f	0.6366	0.0087	0.9871	0.073
H(17B)	8f	0.6857	-0.0682	0.9706	0.073
H(17C)	8f	0.6203	-0.0966	0.9359	0.073
H(18A)	8f	0.7037	0.2279	0.8776	0.070

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Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> <sub>iso</sub>
H(18B)	8 <i>f</i>	0.7262	0.1635	0.9490	0.070
H(18C)	8 <i>f</i>	0.6690	0.2369	0.9349	0.070
H(19A)	8 <i>f</i>	0.6685	0.0805	0.8010	0.086
H(19B)	8 <i>f</i>	0.7271	0.0407	0.8519	0.086
H(19C)	8 <i>f</i>	0.6968	0.0373	0.7883	0.086
H(20A)	8 <i>f</i>	0.6651	0.6888	0.9183	0.076

Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> <sub>iso</sub>
H(20B)	8 <i>f</i>	0.7326	0.6825	0.9282	0.076
H(20C)	8 <i>f</i>	0.7009	0.5863	0.9603	0.076
H(21A)	8 <i>f</i>	0.7588	0.4206	0.8875	0.064
H(21B)	8 <i>f</i>	0.7831	0.5156	0.8461	0.064
H(21C)	8 <i>f</i>	0.7436	0.4163	0.8058	0.064
H(1N)	8 <i>f</i>	0.6378(8)	0.398(2)	0.829(1)	0.019(6)

Table 3. Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> <sub>11</sub>	<i>U</i> <sub>22</sub>	<i>U</i> <sub>33</sub>	<i>U</i> <sub>12</sub>	<i>U</i> <sub>13</sub>	<i>U</i> <sub>23</sub>
C(1)	8 <i>f</i>	0.48790(7)	0.3891(2)	0.83702(9)	0.0239(9)	0.0255(9)	0.0184(9)	0.0006(7)	0.0053(7)	0.0007(7)
C(2)	8 <i>f</i>	0.46576(8)	0.4965(2)	0.8463(1)	0.027(1)	0.028(1)	0.034(1)	0.0039(8)	0.0122(8)	0.0003(8)
C(3)	8 <i>f</i>	0.49928(8)	0.5904(2)	0.8489(1)	0.036(1)	0.023(1)	0.038(1)	0.0054(8)	0.0142(9)	0.0008(8)
C(4)	8 <i>f</i>	0.55455(8)	0.5815(2)	0.8437(1)	0.031(1)	0.023(1)	0.036(1)	0.0039(8)	0.0092(9)	0.0012(8)
C(5)	8 <i>f</i>	0.57503(7)	0.4744(2)	0.83523(9)	0.0222(9)	0.0261(9)	0.023(1)	0.0010(7)	0.0053(7)	0.0020(7)
C(6)	8 <i>f</i>	0.54404(8)	0.0348(2)	0.7931(1)	0.037(1)	0.025(1)	0.024(1)	0.0044(8)	0.0108(8)	0.0002(8)
C(7)	8 <i>f</i>	0.5445(1)	0.0839(2)	0.7930(1)	0.051(1)	0.024(1)	0.036(1)	0.0071(9)	0.001(1)	0.0016(9)
C(8)	4 <i>e</i>	½	0.1411(2)	¾	0.065(2)	0.021(2)	0.039(2)	0	0.003(2)	0
C(9)	8 <i>f</i>	0.5847(1)	0.3515(2)	1.0043(1)	0.062(2)	0.041(1)	0.032(1)	0.004(1)	0.004(1)	0.007(1)
C(10)	8 <i>f</i>	0.6298(1)	0.3189(2)	1.0683(1)	0.047(1)	0.073(2)	0.040(1)	0.005(1)	0.003(1)	0.019(1)
C(11)	8 <i>f</i>	0.6097(1)	0.2057(2)	1.0891(1)	0.062(2)	0.060(2)	0.042(1)	0.025(1)	0.001(1)	0.003(1)
C(12)	8 <i>f</i>	0.5590(1)	0.1748(2)	1.0316(1)	0.066(2)	0.043(1)	0.033(1)	0.009(1)	0.010(1)	0.006(1)
C(13)	8 <i>f</i>	0.41006(9)	0.0998(2)	0.8822(1)	0.041(1)	0.036(1)	0.050(1)	0.0082(9)	0.017(1)	0.007(1)
C(14)	8 <i>f</i>	0.4103(1)	0.3262(2)	0.9542(1)	0.056(1)	0.053(2)	0.041(1)	0.006(1)	0.029(1)	0.004(1)
C(15)	8 <i>f</i>	0.33410(9)	0.2903(2)	0.8128(1)	0.027(1)	0.060(2)	0.060(2)	0.000(1)	0.011(1)	0.014(1)
C(16)	8 <i>f</i>	0.67227(9)	0.6499(2)	0.7663(1)	0.038(1)	0.047(1)	0.048(1)	0.012(1)	0.008(1)	0.008(1)
C(17)	8 <i>f</i>	0.6479(1)	0.0355(2)	0.9516(1)	0.051(1)	0.049(1)	0.041(1)	0.010(1)	0.001(1)	0.013(1)
C(18)	8 <i>f</i>	0.69223(9)	0.1871(2)	0.9141(1)	0.034(1)	0.045(1)	0.056(2)	0.004(1)	0.001(1)	0.000(1)
C(19)	8 <i>f</i>	0.6904(1)	0.0150(2)	0.8232(2)	0.054(2)	0.058(2)	0.066(2)	0.023(1)	0.025(1)	0.001(1)
C(20)	8 <i>f</i>	0.6976(1)	0.6378(2)	0.9215(1)	0.042(1)	0.061(2)	0.045(1)	0.009(1)	0.005(1)	0.015(1)
C(21)	8 <i>f</i>	0.75055(8)	0.4663(2)	0.8457(1)	0.022(1)	0.049(1)	0.056(2)	0.0061(9)	0.010(1)	0.003(1)
N(1)	8 <i>f</i>	0.63088(7)	0.4582(2)	0.83302(9)	0.0243(8)	0.022(1)	0.044(1)	0.0005(7)	0.0099(7)	0.0002(8)
N(2)	8 <i>f</i>	0.54279(6)	0.3803(1)	0.83028(7)	0.0210(7)	0.0220(8)	0.0231(8)	0.0003(6)	0.0055(6)	0.0008(6)
N(3)	8 <i>f</i>	0.46048(6)	0.2887(1)	0.83457(8)	0.0249(8)	0.0245(8)	0.0238(8)	0.0011(6)	0.0069(6)	0.0017(6)
N(4)	8 <i>f</i>	0.58414(7)	0.1058(1)	0.83227(8)	0.0317(8)	0.0223(8)	0.0258(8)	0.0052(7)	0.0048(7)	0.0004(6)
N(5)	4 <i>e</i>	½	0.0910(2)	¾	0.029(1)	0.021(1)	0.022(1)	0	0.0068(9)	0
Si(1)	8 <i>f</i>	0.68647(2)	0.55440(5)	0.84142(3)	0.0237(3)	0.0335(3)	0.0323(3)	0.0060(2)	0.0050(2)	0.0007(2)
Si(2)	8 <i>f</i>	0.40584(2)	0.25630(5)	0.86997(3)	0.0244(3)	0.0310(3)	0.0300(3)	0.0022(2)	0.0101(2)	0.0020(2)
Si(3)	8 <i>f</i>	0.64979(2)	0.05923(5)	0.87760(3)	0.0308(3)	0.0301(3)	0.0331(3)	0.0091(2)	0.0058(2)	0.0024(2)
O(1)	8 <i>f</i>	0.56229(6)	0.2458(1)	0.97516(7)	0.0533(9)	0.0356(8)	0.0256(7)	0.0003(7)	0.0051(6)	0.0008(6)
Li(1)	8 <i>f</i>	0.5397(1)	0.2158(3)	0.8781(2)	0.033(2)	0.028(2)	0.027(2)	0.002(1)	0.008(1)	0.000(1)
Li(2)	8 <i>f</i>	0.4451(1)	0.2322(3)	0.7376(2)	0.031(2)	0.026(2)	0.030(2)	0.003(1)	0.008(1)	0.004(1)

**Acknowledgment.** We thank the Deutsche Forschungsgemeinschaft (SPP1181 "Nanoskalige anorganische Materialien durch molekulares Design") for financial support.

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